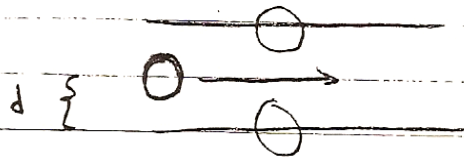


# Collisions

calculate frequency of collisions  $z$



collide with everything inside area  $\pi d^2$

$d$  diameter of molecule

average speed  $\bar{c}$

$\sigma = \pi d^2$  collision cross section

$v'$  volume in which collision could occur = (length)  $\cdot$  (area)  $\cdot$  fraction of collision  
 $\bar{c} \Delta t \cdot \pi d^2 \cdot N \left( \frac{v'}{v} \right)$

divide by  $\Delta t$  get collision frequency

collisions during time period  $\Delta t$   $\rightarrow \bar{c} \Delta t \sigma \frac{N}{V}$   
 collisions per time  $\Delta t$   $\rightarrow \bar{c} \sigma \frac{N}{V}$

$\bar{c}$  should be average relative speed  $\rightarrow \sqrt{2} \bar{c} \sim 1.41 \bar{c}$  because other molecules are moving too also

collision frequency:  $\rightarrow$  for single molecule  $z = \frac{\sqrt{2} \sigma \bar{c} N}{V}$

$\rightarrow$  for  $N$  molecules  $Nz$  but overcount so  $\frac{Nz}{2}$   
 $\rightarrow$  per volume  $V$   $\frac{Nz}{2V}$  since each collision is for 2 molecules

Collisions per unit volume per time  $z_{AA} = \frac{\sigma \bar{c} N^2}{\sqrt{2} V^2}$

$\bar{c} = \left( \frac{8kT}{\pi m} \right)^{1/2}$

$$Z_{AA} = \sigma \left( \frac{N}{V} \right)^2 \left( \frac{4kT}{\pi m} \right)^{1/2}$$

same type of molecules

for two types of molecules

$$Z_{AB} = \pi d_{AB}^2 \left( \frac{8kT}{\pi \mu} \right)^{1/2} \frac{N_A N_B}{V^2}$$

skip

$$\text{where } d_{AB} = \frac{1}{2}(d_A + d_B)$$

properties which may depend on number of collisions imply that possible to calculate molecular diameter

$$d \approx 280 \text{ pm} = 2.8 \text{ \AA} \quad Z \approx 5 \times 10^{34} \text{ s}^{-1} \text{ m}^{-3} \\ \approx 5 \times 10^{28} \text{ s}^{-1} \text{ cm}^{-3}$$

### • Mean Free Path

Calculate mean free path  $\lambda$

molecule at speed  $\bar{c}$  collides frequency  $Z$

time between collisions  $\frac{1}{Z}$

use units  $\rightarrow$

$$\lambda = \left( \frac{\text{m}}{\text{s}} \right) \left( \frac{\text{s}}{\text{collision}} \right) \left( \frac{1}{Z} \right)$$

$$\lambda = \frac{\bar{c}}{Z} = \frac{v(\bar{c})}{\sqrt{2} \sigma \bar{c} N}$$

$$\lambda = \frac{1}{\sqrt{2} \sigma (N/V)}$$

$$\lambda = \frac{1}{\sqrt{2} \sigma (P/kT)}$$

$$\lambda = \frac{kT}{\sqrt{2} \sigma P}$$

$$\lambda = \frac{kT}{\sqrt{2} \sigma P}$$